- 1. Find the de Broglie wavelength λ_b of the following
 - a) An electron whose speed is $10^8\ m/s$

2. An experiment was performed to determine the x-coordinate of an electron with a given wavefunction ψ_0 , by preparing a large assembly of electrons in the same state. This is, all of them have the same wavefunction ψ_0 . In all forty measurements were made, each on a different electron. The results obtained are given below with the number of times the respective values were measured is shown in parenthesis.

-0.08 nm (8), 0.10 nm (4), 0.06 nm (8), 0.08 nm (5), -0.21 nm (1),

-0.12 nm (9), 0.30 nm (2), -0.01 nm (1), 0.35 nm (1), and 0.00 nm (1).

Determine $\langle x \rangle$, $\langle x^2 \rangle$ and hence Δx for the electron. What can be said of the momentum observable?

- 3. A hydrogen atom is 0.5 Angstrom in radius. Use the uncertainty principle to estimate the minimum energy an electron can have in this atom.
- 4. The human eye can detect 5×10^{-18} W of electromagnetic power at the visible wavelengths. How many photons per second does it represent?
- 5. Compton Scattering
 - a) Find the change in the wavelength of 80 pm x-rays that are scattered 120° by a target electron.
 - $(1 \ pm = 10^{-12}m)$
 - b) Find the angle between the directions of the recoil electron and the incident photon.
 - c) Find the energy of the recoil electron
- 6. An electron of mass $m = 10^{-30} kg$ is confined to move along x axis within a distance of 100nm. Determine the energy eigenvalues.
- 7. A particle of mass m is trapped in an infinitely deep square well potential of length L. If it is in the lowest energy state, calculate the probability of being found in
 (a) 0 < x < L/3 and (b) L/3 < x < 2L/3.
- 8. The potential energy for a particle of mass m is given by $V(x) = \frac{1}{2}m\omega^2 x^2$.
 - a) Write the time-independent Schroedinger equation for this case.
 - b) Check if $\psi(x) = Ae^{-x/\alpha}$ can be a wavefunction for the particle with *definite* energy. If yes, find the value of α and the energy.

c) Check if $\psi(x) = Ae^{-x^2/\alpha^2}$ can be a wavefunction for the particle with *definite* energy. If yes, find the value of α and the energy.

b) A grain of sand of mass 0.1mg blown in a wind storm at speeds of 100 km/hr. How would the λ_b compare with the size of the grain.